

# Tetra- $\mu$ -acetato-bis[(9,10-dihydrobenzo[de]imidazo[2,1-a]isoquinolin-7-one- $\kappa$ N<sup>11</sup>)-copper(II)]

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## Key indicators

Single-crystal X-ray study

$T = 298$  K

Mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å

$R$  factor = 0.046

$wR$  factor = 0.124

Data-to-parameter ratio = 15.6

For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

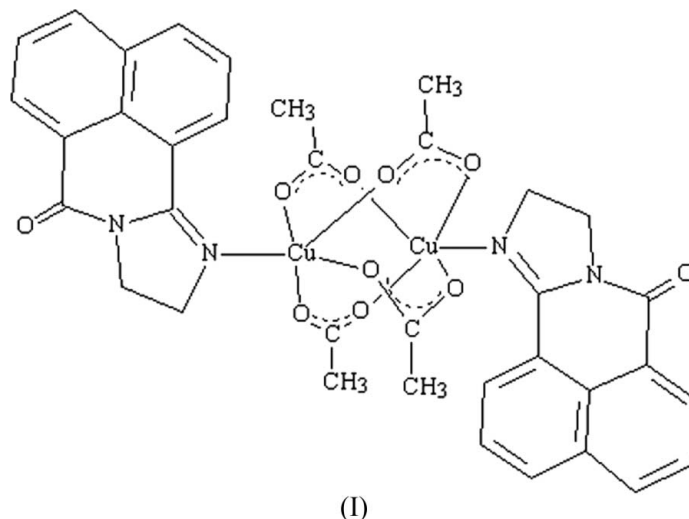
The title dinuclear  $\text{Cu}^{\text{II}}$  complex,  $[\text{Cu}_2(\text{CH}_3\text{COO})_4(\text{C}_{14}\text{H}_{10}\text{N}_2\text{O})_2]$ , is centrosymmetric. The  $\text{Cu}^{\text{II}}$  ion assumes a distorted square-pyramidal coordination geometry. Four acetate groups bridge two  $\text{Cu}^{\text{II}}$  ions, forming a dinuclear molecule.

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## Comment

As part of our ongoing studies on transition metal complexes incorporating imidazole and related ligands (Chen *et al.*, 2005; Yang *et al.*, 2004), the title compound, (I), has been prepared and its X-ray crystal structure is presented here.



The molecular structure of (I) is shown in Fig. 1. The molecule is centrosymmetric. Each  $\text{Cu}^{\text{II}}$  ion is coordinated by a planar ligand of 9,10-dihydro-benzo[de]imidazo[2,1-a]isoquinolin-7-one (*L*) with a longer  $\text{Cu}-\text{N}$  bond distance (Table 1), indicating the Jahn–Teller distortion. Four acetate groups bridge two  $\text{Cu}^{\text{II}}$  ions, forming the dinuclear molecule. Within the dinuclear molecule, the  $\text{Cu}\cdots\text{Cu}$  separation of 2.6809 (9) Å indicates no bonding between  $\text{Cu}^{\text{II}}$  ions. The coordination geometry around the  $\text{Cu}^{\text{II}}$  ion is therefore distorted square-pyramidal.

Weak  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonding occurs between molecules (Table 2).

## Experimental

A methanol solution (10 ml) of *L* (1 mmol) was mixed with an aqueous solution (10 ml) of  $\text{Cu}(\text{CH}_3\text{COO})_2\cdot\text{H}_2\text{O}$  (1 mmol). The mixture was stirred for 1 h at room temperature and then filtered. Single crystals of (I) were obtained from the filtrate after 3 d.

Crystal data

[Cu<sub>2</sub>(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>4</sub>(C<sub>14</sub>H<sub>10</sub>N<sub>2</sub>O)<sub>2</sub>]  
*M<sub>r</sub>* = 807.76  
 Monoclinic, *P*2<sub>1</sub>/*n*  
*a* = 13.2944 (12) Å  
*b* = 9.5110 (8) Å  
*c* = 14.1654 (12) Å  
 $\beta$  = 108.393 (2)°  
*V* = 1699.6 (3) Å<sup>3</sup>

*Z* = 2  
*D<sub>x</sub>* = 1.578 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 $\mu$  = 1.32 mm<sup>-1</sup>  
*T* = 298 (2) K  
 Prism, blue  
 0.17 × 0.14 × 0.09 mm

Data collection

Bruker SMART CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 1998)  
*T<sub>min</sub>* = 0.810, *T<sub>max</sub>* = 0.890

9815 measured reflections  
 3698 independent reflections  
 2465 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.075  
 $\theta_{\max}$  = 27.0°

Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.046  
*wR*(*F*<sup>2</sup>) = 0.124  
*S* = 0.91  
 3698 reflections  
 237 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0576P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.004$   
 $\Delta\rho_{\max} = 0.46 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.38 \text{ e } \text{Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Cu—N1	2.241 (3)	Cu—O4	1.970 (3)
Cu—O2	1.969 (2)	Cu—O5 <sup>i</sup>	1.968 (3)
Cu—O3 <sup>i</sup>	1.970 (2)		
O5 <sup>i</sup> —Cu—O2	88.64 (11)	O4—Cu—O3 <sup>i</sup>	86.96 (12)
O5 <sup>i</sup> —Cu—O4	166.57 (11)	O5 <sup>i</sup> —Cu—N1	99.86 (11)
O2—Cu—O4	91.21 (11)	O2—Cu—N1	100.69 (11)
O5 <sup>i</sup> —Cu—O3 <sup>i</sup>	90.11 (11)	O4—Cu—N1	93.36 (11)
O2—Cu—O3 <sup>i</sup>	166.73 (10)	O3 <sup>i</sup> —Cu—N1	92.54 (11)

Symmetry code: (i)  $-x + 2, -y, -z + 2$ .

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C8—H8...O2	0.93	2.34	3.251 (5)	166
C13—H13...O3 <sup>ii</sup>	0.93	2.51	3.312 (5)	145

Symmetry code: (ii)  $x - 1, y, z$ .

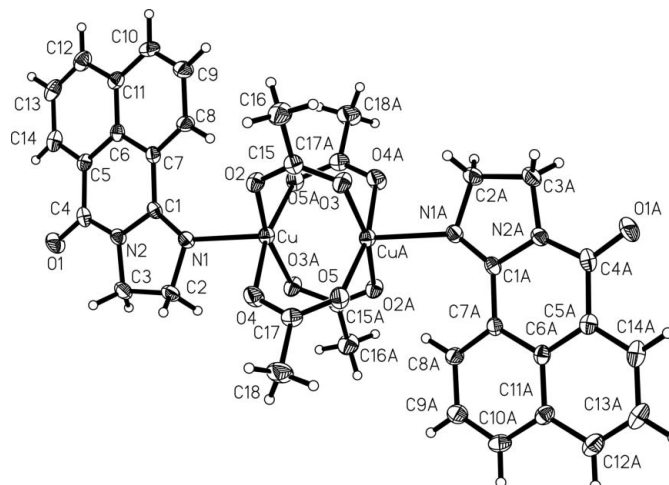


Figure 1

The molecular structure of (I), shown with 30% probability displacement ellipsoids [symmetry code: (A)  $2 - x, -y, 2 - z$ ].

Methyl H atoms were placed in calculated positions, with C—H = 0.96 Å and *U*<sub>iso</sub>(H) = 1.5*U*<sub>eq</sub>(C). Other H atoms were placed in calculated positions, with C—H = 0.93 (aromatic) or 0.97 Å (methylene), and refined in riding mode, with *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(C).

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

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